# SAFETY DATA SHEET

Revision date: 08-Nov-2021



Revision Number 4

# 1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

Product identifier		
Product Name	UREA AMMONIUM NITRATE SOLUTION	
Product Code(s)	00000050287	
Other means of identification		
Recommended use of the chemical and restrictions on use		
Recommended use	Fertiliser.	
Uses advised against	No information available.	

Supplier Ixom Operations Pty Ltd ABN: 51 600 546 512 Level 8, 1 Nicholson Street Melbourne 3000 Australia

Telephone Number: +61 3 9906 3000

# Emergency telephone number

Emergency telephone number

# 1 800 033 111 (ALL HOURS)

Please ensure you refer to the limitations of this Safety Data Sheet as set out in the "Other Information" section at the end of this Data Sheet.

# 2. HAZARDS IDENTIFICATION

# GHS Classification

Not classified as dangerous goods in accordance with the Australian Code for the Transport of Dangerous Goods by Road and Rail (ADG)

Classified as a hazardous chemical in accordance with the criteria of Safe Work Australia - Globally Harmonized System (GHS).

Category 2

SIGNAL WORD Warning

Label elements

Exclamation mark



Hazard statements H319 - Causes serious eye irritation

#### **Precautionary Statements - Prevention**

Wash hands and face thoroughly after handling
Wear eye protection/ face protection
IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing If eye irritation persists: Get medical advice/attention
Precautionary Statements - Storage
No storage statements
Precautionary Statements - Disposal
No disposal statements.

Other hazards which do not result in classification General Hazards

Poisons Schedule (SUSMP) None allocated

# 3. COMPOSITION/INFORMATION ON INGREDIENTS

Mixture

**Chemical nature** 

Urea ammonium nitrate solutions with 28-32% w/w N.

Chemical name	CAS No.	Weight-%
Ammonium nitrate	6484-52-2	40-50
Urea	57-13-6	30-40
Water and other non-hazardous components	-	<30
Ammonia	7664-41-7	0.1

# 4. FIRST AID MEASURES

### **Description of first aid measures**

General advice	Show this safety data sheet to the doctor in attendance.
Emergency telephone number	Poisons Information Center, Australia: 13 11 26 Poisons Information Center, New Zealand: 0800 764 766
Inhalation	Remove to fresh air and keep at rest in a position comfortable for breathing. If breathing is difficult, (trained personnel should) give oxygen. Get medical attention immediately if symptoms occur.
Eye contact	Rinse immediately with plenty of water, also under the eyelids, for at least 15 minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Keep eye wide open while rinsing. Get medical attention if irritation develops and persists.
Skin contact	Wash off immediately with plenty of water. Get medical attention if irritation develops and persists. Take off contaminated clothing and wash before reuse.
Ingestion	Rinse mouth immediately and drink plenty of water. Do NOT induce vomiting. Drink 1 or 2

	glasses of water. Get medical attention. Never give anything by mouth to an unconscious person. If victim has breathing difficulties treat as for "Inhalation".	
Self-protection of the first aider	Avoid contact with skin, eyes, and clothing. Ensure that medical personnel are aware of the material(s) involved, take precautions to protect themselves and prevent spread of contamination. Use personal protective equipment as required. See section 8 for more information.	
Most important symptoms and effect	cts, both acute and delayed	
Symptoms	Irritation. May cause redness and tearing of the eyes.	
Indication of any immediate medica	I attention and special treatment needed	
Note to physicians	<ul> <li>Treat symptomatically. The absorption of this product into the body may lead to the formation of methemoglobin that, in sufficient concentration, causes cyanosis. Cyanosis is clinically detectable when approximately 15% of the haemoglobin has been converted to methaemoglobin (ferric iron).</li> <li>Clinical findings: The smooth muscle relaxant effect of nitrate salts may lead to headache, dizziness and marked hypotension. Symptoms such as headache, dizziness, weakness and dyspnoea occur when methaemoglobin concentrations are 30% to 40%; at levels of about 60%, stupor, convulsions, coma and respiratory paralysis occur and the blood is a chocolate brown colour. At higher levels death may result. Spectrophotometric analysis can determine the presence and concentration of methaemoglobin in blood.</li> <li>Treatment:</li> <li>Give 100% oxygen.</li> <li>In cases of (a) ingestion: use gastric lavage, (b) contamination of skin (unburnt or burnt): continue washing to remove salts.</li> <li>Observe blood pressure and treat hypotension if necessary.</li> <li>When methaemoglobin concentrations exceed 40% or when symptoms are present, give methylene blue 1 to 2mg/kg body weight in a 1% solution by slow intravenous injection. If cyanosis has not resolved within one hour a second dose of 2 mg/kg body weight may be given. The total dose should not exceed 7 mg/kg body weight as unwanted effects such as dyspnoea, chest pain, vomiting, diarrhoea, mental confusion and cyanosis may occur. Without treatment methaemoglobin levels in excess of 40%.</li> <li>Continue to monitor and give oxygen for at least two hours after treatment with methylene blue.</li> <li>Consider transfer to centre where haemoperfusion can be performed to remove the nitrates from the blood if the condition of the patient is unstable.</li> <li>Following inhalation of oxides of nitrogen the patient should be observed in hospital for 24 hours for delayed on setof pulmonary oedema.</li> </ul>	
5. FIRE FIGHTING MEASU	RES	
Suitable Extinguishing Media		
Suitable Extinguishing Media	Use extinguishing agent suitable for type of surrounding fire. Water spray or fog. Cool containers with flooding quantities of water until well after fire is out.	
Unsuitable extinguishing media	Carbon dioxide (CO2). Dry chemical. Do not use extinguishing media that contains ammonium salts.	
Specific hazards arising from the chemical		
Specific hazards arising from the	Weak oxidiser. Non-combustible, substance itself does not burn but may decompose upon	

chemical	heating to produce corrosive and/or toxic fumes. Do not allow evaporation to dryness. Nitrate salts on their own are not combustible, however, they will support the combustion of other materials. Decomposes on heating emitting irritating white fumes and/or brown fumes. Brown fumes indicate the presence of toxic oxides of nitrogen. On detection of fire the compartment(s) should be opened up to provide maximum ventilation. Fire-fighters to wear self-contained breathing apparatus and suitable protective clothing if there is a risk of exposure to products of combustion/decomposition. Fires should be fought from a protected location. Keep containers and adjacent areas cool with water spray. If safe to do so, remove containers from path of fire. If safe to do so, prevent molten material from being confined in drains, pipes etc. A major fire may involve a risk of explosion. An adjacent detonation may also involve the risk of explosion.	
Hazardous combustion products	Carbon oxides. Nitrogen oxides. Nitric acid. Ammonia.	
Special protective actions for fire-fighters		
Special protective equipment for fire-fighters	Firefighters should wear self-contained breathing apparatus and full firefighting turnout gear. Move containers from fire area if you can do it without risk. Cool containers with	

flooding quantities of water until well after fire is out.

# 6. ACCIDENTAL RELEASE MEASURES

# Personal precautions, protective equipment and emergency procedures

Personal precautions	Evacuate personnel to safe areas. Remove all sources of ignition. Avoid contact with skin and eyes. Ensure adequate ventilation. Wear protective gloves/protective clothing and eye/face protection. Do not touch or walk through spilled material.	
Other information	Refer to protective measures listed in Sections 7 and 8.	
For emergency responders	Use personal protection recommended in Section 8.	
Environmental precautions		
Environmental precautions	Prevent further leakage or spillage if safe to do so. Keep out of waterways. Local authorities should be advised if significant spillages cannot be contained.	
Methods and material for containment and cleaning up		
Methods for containment	Contain and collect spillage with non-combustible absorbent material, (e.g. sand, earth, diatomaceous earth, vermiculite) and place in container for disposal according to local / national regulations (see Section 13). Dike far ahead of spill to collect runoff water.	
Methods for cleaning up	Use a non-combustible material like vermiculite, sand or earth to soak up the product and place into a container for later disposal. Collect in properly labelled drums or other suitable containers, with loose fitting lids. Never return spill or leaks to original containers for re-use.	

# 7. HANDLING AND STORAGE

Precautions for safe handling	
Advice on safe handling	Handle in accordance with good industrial hygiene and safety practice. Avoid contact with skin and eyes. Wash thoroughly after handling.
General hygiene considerations	Handle in accordance with good industrial hygiene and safety practice. Do not eat, drink or smoke when using this product. Take off contaminated clothing and wash it before reuse. Wear suitable gloves and eye/face protection. Do not get in eyes, on skin, or on clothing.
Conditions for only storens include	

# Conditions for safe storage, including any incompatibilities

Storage Conditions	Store in a cool, dry area away from potential sources of heat, open flames, sunlight or other chemicals. Keep container closed when not in use.
Incompatible materials	Incompatible with permanganates. Alkalis. Reducing agents. Finely powdered metals. Combustible material. Nitrite. Chlorites. Chlorine. Chlorides. Strong acids. Copper. Zinc. Brass. Bronze.
Poisons Schedule (SUSMP)	None allocated

# 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

### **Control parameters**

Exposure Limits

No value assigned for this specific material by Safe Work Australia. However, Workplace Exposure Standard(s) for constituent(s) and decomposition product(s):

Ammonia: 8hr TWA = 17 mg/m<sup>3</sup> (25 ppm), 15 min STEL = 24 mg/m<sup>3</sup> (35 ppm) Nitrogen dioxide: 8hr TWA = 5.6 mg/m<sup>3</sup> (3 ppm), 15 min STEL = 9.4 mg/m<sup>3</sup> (5 ppm)

As published by Safe Work Australia Workplace Exposure Standards for Airborne Contaminants.

TWA - The time-weighted average airborne concentration of a particular substance when calculated over an eight-hour working day, for a five-day working week.

STEL (Short Term Exposure Limit) - the airborne concentration of a particular substance calculated as a time-weighted average over 15 minutes, which should not be exceeded at any time during a normal eight hour work day. According to current knowledge this concentration should neither impair the health of, nor cause undue discomfort to, nearly all workers.

These Workplace Exposure Standards are guides to be used in the control of occupational health hazards. All atmospheric contamination should be kept to as low a level as is workable. These workplace exposure standards should not be used as fine dividing lines between safe and dangerous concentrations of chemicals. They are not a measure of relative toxicity.

### Appropriate engineering controls

 Engineering controls
 Apply technical measures to comply with the occupational exposure limits.

 If in the handling and application of this material, safe exposure levels could be exceeded, the use of engineering controls such as local exhaust ventilation must be considered and the results documented. If achieving safe exposure levels does not require engineering controls, then a detailed and documented risk assessment using the relevant Personal Protective Equipment (PPE) (refer to PPE section below) as a basis must be carried out to determine the minimum PPE requirements.

### Individual protection measures, such as personal protective equipment

The selection of PPE is dependent on a detailed risk assessment. The risk assessment should consider the work situation, the physical form of the chemical, the handling methods, and environmental factors.

OVERALLS, SAFETY SHOES, CHEMICAL GOGGLES, GLOVES.



# 9. PHYSICAL AND CHEMICAL PROPERTIES

Physical state	Liquid	
Appearance	No information available.	
Color	Pale Blue	
Odor	Slight Ammonia	
Odor threshold	No information available.	
Property	Values	Remarks • Method
рН	6.5-7.5	
pH (as aqueous solution)	No data available	None known
Melting point / freezing point	No data available	None known
Boiling point / boiling range	>100°C	
Flash point	Not applicable	None known
Evaporation rate	No data available	None known
Flammability (solid, gas)	No data available	None known
Flammability Limit in Air		None known
Upper flammability or explosive limits	No data available	
Lower flammability or explosive limits	No data available	
Vapor pressure	No data available	None known
Vapor density	No data available	None known
Relative density	1.28-1.32	
Water solubility	Miscible in water	None known
Solubility(ies)	No data available	None known
Partition coefficient	No data available	None known
Autoignition temperature	No data available	None known
Decomposition temperature	No data available	None known
Kinematic viscosity	No data available	None known

No data available

Other information

Dynamic viscosity

# **10. STABILITY AND REACTIVITY**

Reactivity

None known

Reactivity	Weak oxidiser.
Chemical stability	
Stability	Stable under normal conditions.
Explosion data Sensitivity to mechanical impac	<b>t</b> None.
Sensitivity to static discharge	None.
Possibility of hazardous reactions	
Possibility of hazardous reactions	None under normal processing.
Hazardous polymerization	Hazardous polymerization does not occur.
Conditions to avoid	
Conditions to avoid	Avoid contact with combustible substances. Do not allow evaporation to dryness. Avoid contact with other chemicals. Avoid contamination of the material. Keep away from open flames, hot surfaces and sources of ignition.
Incompatible materials	
Incompatible materials	Incompatible with permanganates. Alkalis. Reducing agents. Finely powdered metals. Combustible material. Nitrite. Chlorites. Chlorine. Chlorides. Strong acids. Copper. Zinc. Brass. Bronze.

Hazardous decomposition products

Hazardous decomposition products Carbon oxides. Nitrogen oxides. Ammonia. Nitric acid. Oxygen.

# **11. TOXICOLOGICAL INFORMATION**

# Acute toxicity

Information on likely routes of exposure

Product Information	No adverse health effects expected if the chemical is handled in accordance with this Safety Data Sheet and the chemical label. Symptoms or effects that may arise if the chemical is mishandled and overexposure occurs are:
Inhalation	Inhalation of aerosols:. May cause irritation of respiratory tract.
	Absorption of ammonium nitrate by inhalation, ingestion or through burnt or broken skin may cause dilation of blood vessels by direct smooth muscle relaxation and may also cause methaemoglobinaemia. May cause dizziness, drowsiness, nausea and headache due to central nervous system effects.
Eye contact	Causes serious eye irritation.
Skin contact	May cause irritation. The ammonium nitrate component of this material can be absorbed through burnt, cut or broken skin with resultant adverse effects. See effects as noted under 'Inhalation'.
Ingestion	Ingestion may cause gastrointestinal irritation, nausea, vomiting and diarrhoea. May cause a lowering of blood pressure (hypotension). Ingestion of larger amounts may cause defects to the central nervous system (e.g. dizziness, headache).
Symptoms	Irritation. May cause redness and tearing of the eyes.

# Numerical measures of toxicity - Product Information

No information available.

# **Component Information**

	Chemical name Oral LD50		Dermal LD50	Inhalation LC50		
Ammonium nitrate = 2217 mg/kg (Rat)		-	> 88.8 mg/L (Rat)4 h			
	Urea	= 8471 mg/kg (Rat)	-	-		
	Ammonia	= 350 mg/kg (Rat)	-	= 2000 ppm (Rat)4 h		

See section 16 for terms and abbreviations

### Delayed and immediate effects as well as chronic effects from short and long-term exposure

Skin corrosion/irritation	No information available.		
Serious eye damage/eye irritation	Causes serious eye irritation. Classification is based on mixture calculation methods based on component data.		
Respiratory or skin sensitization	No information available.		
Germ cell mutagenicity	No information available.		
Carcinogenicity	No information available.		
Reproductive toxicity	No information available.		
STOT - single exposure	No information available.		
STOT - repeated exposure	No information available.		
Aspiration hazard	No information available.		
Chronic effects:	In humans and animals methaemoglobinaemia has occurred under untreated circumstances following overexposure to nitrates. Absorption of nitrates by any route may cause dilation of blood vessels by direct smooth muscle relaxation. May cause anaemia and methemoglobinaemia, characterised by dizziness, drowsiness,		
	headache, breath shortness, cyanosis (bluish skin due to deficient oxygenation of the		

# **12. ECOLOGICAL INFORMATION**

# **Ecotoxicity**

ECOLOXICITY					
Ecotoxicity	otoxicity Keep out of waterways.				
		Ammonium nitrate is a plant nutrient. Large scale contamination may kill vegetation and cause poisoning in livestock and poultry.			
	magna was d	Ammonium nitrate was evaluated at 5, 10, 25 and 50 mg (NH4+)/L. The fertility of Daphnia magna was decreased at 50 mg/L. Post embryonic growth of crustacea was impaired at 10, 25 and 50 mg/L. Can stimulate weed and algal growth in static surface waters.			
Chemical name	Algae/aquatic plants	Fish	Toxicity to	Crustacea	

blood), rapid heart rate and chocolate-brown coloured blood.

			microorganisms	
Ammonium nitrate	-	LC50: 65 - 85mg/L (48h,	-	-
		Cyprinus carpio)		
Urea	-	LC50: 16200 -	-	EC50: =3910mg/L (48h,
		18300mg/L (96h, Poecilia		Daphnia magna) EC50:
		reticulata)		>10000mg/L (24h,
				Daphnia magna Straus)
Ammonia	-	LC50: =0.44mg/L (96h,	-	LC50: =25.4mg/L (48h,
		Cyprinus carpio) LC50:		Daphnia magna)
		0.26 - 4.6mg/L (96h,		
		Lepomis macrochirus)		
		LC50: =1.17mg/L (96h,		
		Lepomis macrochirus)		
		LC50: 0.73 - 2.35mg/L		
		(96h, Pimephales		
		promelas) LC50:		
		=5.9mg/L (96h,		
		Pimephales promelas)		
		LC50: >1.5mg/L (96h,		
		Poecilia reticulata) LC50:		
		=1.19mg/L (96h, Poecilia		
		reticulata)		

# Persistence and degradability

Persistence and degradability No information available.

# Bioaccumulative potential

Bioaccumulation

No information available.

#### **Component Information**

Chemical name	Partition coefficient	
Ammonium nitrate	-3.1	
Urea	-1.59	
Ammonia	-1.14	

<u>Mobility</u>

Mobility in soil No information available.

Other adverse effects

Other adverse effects

Ammonium nitrate was evaluated at 5, 10, 25 and 50 mg (NH4+)/L. The fertility of Daphnia magna was decreased at 50 mg/L Post embryonic growth of crustacea was impaired at 10, 25 and 50 mg/L.

# **13. DISPOSAL CONSIDERATIONS**

# Waste treatment methods

Waste from residues/unused products	Dispose of in accordance with local regulations. Dispose of waste in accordance with environmental legislation.
Contaminated packaging	Empty containers should be taken to an approved waste handling site for recycling or disposal.

# 14. TRANSPORT INFORMATION

### ADG

Not classified as Dangerous Goods by the criteria of the Australian Dangerous Goods Code (ADG Code) for transport by Road and Rail; NON-DANGEROUS GOODS.

# <u>IATA</u>

Not classified as Dangerous Goods by the criteria of the International Air Transport Association (IATA) Dangerous Goods Regulations for transport by air; NON-DANGEROUS GOODS.

#### IMDG

Not classified as Dangerous Goods by the criteria of the International Maritime Dangerous Goods Code (IMDG Code) for transport by sea; NON-DANGEROUS GOODS.

# **15. REGULATORY INFORMATION**

### Safety, health and environmental regulations/legislation specific for the substance or mixture

### National regulations

#### Australia

Not classified as dangerous goods in accordance with the Australian Code for the Transport of Dangerous Goods by Road and Rail (ADG)

Classified as a hazardous chemical in accordance with the criteria of Safe Work Australia - Globally Harmonized System (GHS).

See section 8 for national exposure control parameters

#### Poisons Schedule (SUSMP) None allocated

# Major hazard (accident/incident planning) regulation

Verify that license requirements are met

Chemical name	Threshold quantity (T)	
Ammonia - 7664-41-7	200 tonne TQ anhydrous, liquefied or solution; relative density	
	<0.880 at 15°C in water; with >50% Ammonia	

# National pollutant inventory

Subject to reporting requirement		
Chemical name	National pollutant inventory	
Urea - 57-13-6	20 MW Threshold category 2b total	
	60000 MWH Threshold category 2b total	
	1 tonne/h Threshold category 2a total	
	25 tonne/yr Threshold category 1a total	
	400 tonne/yr Threshold category 2a total	
	2000 tonne/yr Threshold category 2b total	
Ammonia - 7664-41-7	10 tonne/yr Threshold category 1 total	

# International Inventories

AIIC

All the constituents of this material are listed on the Australian Inventory of Industrial Chemicals.

#### Legend:

- Australian Inventory of Industrial Chemicals

### International Regulations

The Montreal Protocol on Substances that Deplete the Ozone Layer Not applicable

The Stockholm Convention on Persistent Organic Pollutants Not applicable

The Rotterdam Convention Not applicable

# **16. OTHER INFORMATION**

Supplier Safety Data Sheet 11/2017

**Reason(s) For Issue:** 5 Yearly Revised Primary SDS

Issuing Date: 08-Nov-2021

This Safety Data Sheet has been prepared by Ixom Operations Pty Ltd (Toxicology and SDS Services).

#### **Revision Note:**

The symbol (\*) in the margin of this SDS indicates that this line has been revised.

### Key or legend to abbreviations and acronyms used in the safety data sheet

Legend Section	on 8: EXPOSURE CONTROLS/PERSONAL	PROTECTION	
TWA	TWA (time-weighted average)	STEL	STEL (Short Term Exposure Limit)
Ceiling	Maximum limit value	*	Skin designation
С	Carcinogen		C C

#### Key literature references and sources for data used to compile the SDS

EPA (Environmental Protection Agency) Acute Exposure Guideline Level(s) (AEGL(s)) U.S. Environmental Protection Agency Federal Insecticide, Fungicide, and Rodenticide Act U.S. Environmental Protection Agency High Production Volume Chemicals Food Research Journal Hazardous Substance Database International Uniform Chemical Information Database (IUCLID) Japan GHS Classification Australian Industrial Chemicals Introduction Scheme (AICIS) NIOSH (National Institute for Occupational Safety and Health) National Library of Medicine's ChemID Plus (NLM CIP) National Library of Medicine's PubMed database (NLM PUBMED) National Toxicology Program (NTP) New Zealand's Chemical Classification and Information Database (CCID) Organization for Economic Co-operation and Development Environment, Health, and Safety Publications Organization for Economic Co-operation and Development High Production Volume Chemicals Program Organization for Economic Co-operation and Development Screening Information Data Set RTECS (Registry of Toxic Effects of Chemical Substances) World Health Organization

### **Disclaimer**

This SDS summarises to our best knowledge at the date of issue, the chemical health and safety hazards of the material and general guidance on how to safely handle the material in the workplace. Since Ixom Operations Pty Ltd cannot anticipate or control the conditions under which the product may be used, each user must, prior to usage, assess and control the risks arising from its use of the material.

If clarification or further information is needed, the user should contact their Ixom representative or Ixom Operations Pty Ltd at the contact details on page 1.

Ixom Operations Pty Ltd's responsibility for the material as sold is subject to the terms and conditions of sale, a copy of which is available upon request.

End of Safety Data Sheet